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## **Balancing Chemical Equations**

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## **Gizmo Worksheet**

The equation is balanced when there are equal numbers of each type of atom represented on each side of the equation. In the Gizmo, use the up and down arrows to adjust the numbers of hydrogen, oxygen, and water molecules until the equation is balanced. When you are done, turn on

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Balance and classify five types of chemical reactions: synthesis, decomposition, single replacement, double replacement, and combustion. While balancing the reactions, the number of atoms on each side is presented as visual, histogram, and

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numerical data.

## **Balancing Chemical Equations Gizmo : ExploreLearning**

This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction:  $\text{H}_2 + \text{O}_2 \Rightarrow \text{H}_2\text{O}$ . 1. How

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many atoms are in a hydrogen molecule ( $H_2$ )? 2 2. How many atoms are in an oxygen molecule ( $O_2$ )? 2 3.

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Balancing Equations Gizmo Worksheets - there are 8 printable worksheets for this topic. Worksheets are Balancing



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Therefore, a balanced chemical equation will show the same number of each type of atom on each side of the equation. To set up an equation in the Chemical Equations Gizmo, type the chemical...

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number of atoms on each side is presented as visual, histogram, and numerical data. Video Preview Launch Gizmo.

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## **Gizmo Ionic Worksheets - Kiddy Math**

When you find difficulty in balancing the equation in the balancing chemical equations worksheet, you can miss it with a fraction of  $\frac{1}{2}$  and that will easily balance the equation. But the problem is that you cannot have a fraction for the

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co-efficient, this is why doubling all coefficients will help you balance the equation.

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This is a collection of printable worksheets to practice balancing equations. The printable worksheets are provided with separate answer keys.

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Equations - Worksheet #2 Balancing

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## **Balancing Chemical Equations Worksheet Answer Key ...**

Balance the following chemical equations.

1.  $\text{Fe} + \text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + \text{H}_2$
2.  $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{CO}_2$
3.  $\text{KOH} + \text{H}_3\text{PO}_4 \rightarrow \text{K}_3\text{PO}_4 + \text{H}_2\text{O}$
4.  $\text{SnO}_2 + \text{H}_2 \rightarrow \text{Sn} + \text{H}_2\text{O}$
5.  $\text{NH}_3 + \text{O}_2 \rightarrow \text{NO}$

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+ H<sub>2</sub>O : 6. KNO<sub>3</sub> + H<sub>2</sub>CO<sub>3</sub>: K<sub>2</sub>CO<sub>3</sub> +  
HNO<sub>3</sub>: 7. B<sub>2</sub>Br<sub>6</sub> + HNO<sub>3</sub>: B(NO<sub>3</sub>)<sub>3</sub> +  
HBr : 8. 3 + BF<sub>3</sub> Li<sub>2</sub>SO<sub>3</sub>: B<sub>2</sub>(SO<sub>3</sub>)<sub>3</sub> +  
LiF : 9. (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> + Pb(NO<sub>3</sub>)<sub>4</sub>: Pb ...

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## **Balancing Chemical Equations Answer Sheet Worksheets ...**

A chemical equation describes what happens in a chemical reaction. The equation identifies the reactants (starting materials) and products (resulting substances), the formulas of the participants, the phases of the



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participants (solid, liquid, gas), the direction of the chemical reaction, and the amount of each substance. Chemical equations are balanced for mass and charge, meaning the number ...

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